

Answers To Explore Learning Calorimetry Lab Gizmo

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Investigate how calorimetry can be used to find relative specific heat values when different substances are mixed with water. Modify initial mass and temperature values to see effects on the system. One or any combination of the substances can be mixed with water. A dynamic graph (temperature vs. time) shows temperatures of the individual substances after mixing.

Calorimetry Lab Gizmo--Lesson Info--Explore Learning

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Explore Learning Calorimetry Lab Answer Key

water. Calorimetry Lab Answers Correct Answer: A. Substance A A chemist mixes 500 g of lead at 500°C with 1,200 g of water at 20°C. She then mixes 500 g of copper at 500°C with 1,200 g of water at 20°C. The specific heat capacity of lead is 0.1276 J/g°C and the specific heat capacity of copper is 0.3845 J/g°C.

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