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1. Write the chemical and oxidation number of the positive ion. If the positive ion is monoatomic (only one atom), you can find its oxidation number from the periodic table. If the positive ion is polyatomic, use

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element valence electrons  
number needed element valence  
electrons number needed H11  
Ne80 He20 Na17 Li17 Mg26 Be26

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Al35 B35 Si44 C44 P53 N53 S62  
O62 Cl71 F71 Ar80 Figure 19.3:  
Helium atoms have only two  
electrons, both of which are in the  
outermost level. Helium is an  
exception to the octet rule.

Changes in Matter Chapter 19

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### Molecules and

19.2 Chemical Formulas  $\square\square$  ions an atom that has an electrical charge. ionic compound A compound made out of ions. Oxidation number Shows how many electrons are lost, Gained or

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