

Chapter 12 Stoichiometry Guided Reading Study Work

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SECTION 12.1 THE ARITHMETIC OF EQUATIONS

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Chapter 12 Guided Reading Stoichiometry Answer Key

Chapter 12 Stoichiometry Guided Reading Chapter 12 Stoichiometry 127. SECTION 12.1 THE ARITHMETIC OF EQUATIONS (pages 353 – 358) This section explains how to calculate the amount of reactants required or product formed in a nonchemical process.

Chapter 12 Stoichiometry Guided Reading Answers

Introduce the term sto- ichiometry in your own words. Stress that stoichiometry allows students to calculate the amounts of chemical sub- stances involved in chemical reactions using information obtained from bal- anced chemical equations.

12.1 The Arithmetic of Equations 12

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Chapter 3: Stoichiometry – Guided Reading Section 3.1 – 3.2 1. True or False? Most hydrogen atoms have a mass of 1.008 amu. Justify your answer. If true, explain why it is true. If false, what mass do most hydrogen atoms have? False, 1.008 amu is actually hydrogen ' s average mass, NO atom of hydrogen actually has the mass of 1.008 amu. 2.

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