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18-1 CHAPTER 18 ACID-BASE EQUILIBRIA Section 18.1: Acids and Bases in Water Water (H₂O) - the most important molecule on earth. Even in pure water, there are small amounts of ions from the equilibrium below ("self-ionization of water" or "auto-ionization).

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model of acids and bases states that an acid contains the element hydrogen and forms ions of this element when it is dissolved in water. a base contains the hydroxide group and dissociates to produce hydroxide ions in aqueous solution

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Optional study guide: Chemistry: Matter and Its Changes, Study Guide, 4th Edition, by James E. Brady, Fred Senese, Published by Wiley ... Chapter 18: Equilibrium in Solutions of Weak Acids and Bases (sections 1-8).

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Chapter 18 Study Guide (Acids and Bases) STUDY. PLAY. A neutral solution contains more hydroxide ions than hydrogen ions. False. An aqueous solution of acid causes red litmus paper to turn blue. False. In a typical reaction, if the reacting acid is strong, the base produced is strong.

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Acids are named by changing the ending of the alkane to -oic acid. i.e. a 4 carbon acid goes from butane to butanoic acid Acid salts are named by changing the acid ending to -oate i.e the 4 carbon acid goes from butanoic acid to sodium butanoate Esters are named by changing the ending of the alkane to -oate and by using alkyl name i.e. the 4 carbon ester below is methyl butanoate (methyl for OCH₃ portion) O O Anhydrides are dehydrated esters that replace -oic acid with word anhydride Acyl ...

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Chap 18 Acid Bases Study Guide Answers 162 Chapter 5 Acids, Bases, and Acid-Base Reactions. acid, H_3PO_4 , are triprotic acids. Most of the phosphoric acid produced by the chemical industry is used to make fertilizers and detergents, but it is also used to make pharmaceuticals,

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