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law of thermodynamics becomes: $U + A = 0$; (1) where A is the work done by gas, and U is its internal energy. Using $A = P_2 V_2 - P_1 V_1$ and $U = C_V (T_2 - T_1) + P_2 (V_2 - V_1) = 0$; (2)

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