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*Worked Chemistry Problem Examples*

Saturated ionic solutions may be used to decrease the solubility of organic compounds in the aqueous layer, allowing more of a compound to dissolve in the organic layer. If a desired product can hydrogen bond with water and is relatively small, it may be difficult to keep it in the organic layer when partitioning with an aqueous phase ( $(K)$  will be  $<1$ ).

*4.7: Reaction Work-Ups - Chemistry LibreTexts*

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